

The Crystal and Molecular Structure
Of Meta-Aminobenzoic Acid Hydrochloride

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ABSTRACT

The structure of crystals of meta-aminobenzoic acid hydrochloride has been determined by two two-dimensional Patterson projections and refined by three-dimensional least-squares techniques. The crystals are triclinic, space group $P\bar{1}$, with unit cell dimensions: $a=4.562$, $b=5.823$, $c=14.724$, $\alpha=98^{\circ}39'$, $\beta=91^{\circ}12'$, $\gamma=95^{\circ}45'$. There are two molecules in the unit cell. The R index of the refined structure is 11.26% with 1507 reflections. Pairs of molecules are linked together to form dimers through two O-H...O hydrogen bonds arranged about a center of symmetry; and the amino and benzene ring fall in the same plane, with the oxygens in the carboxyl groups tilted out of the plane by 0.12 angstroms.

INTRODUCTION

Amino acids occur in certain biological processes which makes them of some interest to the investigation of molecular biology. As a continuing project regarding the structure of amino acids in this lab, we have undertaken the investigation of meta-aminobenzoic acid because of its biological interest and for the sake of comparing the meta- to the para- form which has previously been determined and reported in the literature. (Ref. 1)

Meta-aminobenzoic acid is a stable gray powder. Attempts to crystallize the pure compound from common inorganic and organic solvents proved to be unsuccessful; crystals of the hydrochloride form were, however, prepared and used for this investigation. The hydrochloride was formed by dissolving the compound in hot, concentrated hydrochloric acid. Then the solution was allowed to slowly cool, crystallizing out the hydrochloride. The crystals were then left to dry in the air. The crystals were long needles of a few millimeters in length with some twinning and cracking present. The crystals were found to be very soluble in water, but they were stable for long periods of time in air.

EXPERIMENTAL

Two crystals were used for the determination of the unit cell dimensions and for the collection of intensity data, one aligned with the a axis and the other with the b axis, parallel to the rotation axes. Unit cell dimensions were determined from zero-level precision Weissenberg photographs along the axes of both crystals. From these, the three reciprocal space lattice distances (a^* , b^* , c^*) and the two reciprocal space lattice angles (α^* , β^*) were determined. The third reciprocal space lattice angle (γ^*) was taken from an $hk0$ precession photograph. The cell dimensions were then calculated and are summarized in the subsequent table:

$a = 4.562 \text{ \AA.}$	$\sigma = 0.0004$
$b = 5.823 \text{ \AA.}$	$\sigma = 0.0006$
$c = 14.724 \text{ \AA.}$	$\sigma = 0.0011$
$\alpha = 98^\circ 39'$	$\sigma = 0.7'$
$\beta = 91^\circ 12'$	$\sigma = 0.2'$
$\gamma = 95^\circ 45'$	$\sigma = 1'$

The molecule forms a triclinic lattice, space group $P\bar{1}$, with two molecules in the unit cell being related by an inversion center.

Intensities were estimated visually from multiple-film equi-inclination Weissenberg photographs using $\text{CuK}\alpha$ radiation. Films were taken of layers 0 to 3 about the a axis and layers 0 to 4 about the b axis. A standard deviation was assigned to each reflection according to an expression given by Sharma and McConnell (Ref. 2) and was used in obtaining film and pack factors for each set of photographs. These standard deviations were also used for determining an average intensity and assigning a standard deviation to the reflections. These and all subsequent calculations used the CRYRM system (Ref. 3) on an IBM 7094-7040 computer. Intensities were corrected for Lorentz-polarization but not for absorption. The resulting list of reflections (Table 4) was comprised of 1507 reflections, of which 248 were too weak to be observed above the background.

The heavy atom parameters were first determined from zero-layer data about the two axes. Two-dimensional Patterson maps were made, and the chlorine coordinates were taken from these Pattersons. The coordinates of the remaining heavy atoms were found using a Fourier map calculated from the phases ^d based upon the chlorine coordinates.

These trial parameters were then used in a structure-factor and least-squares cycles for the complete three-dimensional data. The structure was refined to an R factor of 11.8% with the addition of the hydrogen atoms in their expected positions.

The intensities of 290 reflections whose observed and calculated structure factors showed marked disagreement were re-evaluated. Re-reading of intensities and changes in the weighting scheme were used to correct these reflections. Refinement of the structure then yielded an R factor of 11.26% with the hydrogen atoms not included in the refinement. Further evaluation of the data is still in progress.

Fourier (Fig. 1) and difference Fourier maps show the existence of no other heavy atoms in the unit cell. Tables 1, 2, and 3, demonstrate the results of the least-squares refinement.

CONCLUSIONS

An interesting feature of the structure arises from the chlorine to chlorine distance across the *b* and *c* axes. These distances are 3.82 and 3.60 angstroms respectively. This configuration is unfavorable by the chloride-chloride coulomb repulsion; however, this is balanced by the chloride-amine coulomb attractions. Another facet of the structure which is a common feature with carboxyl groups is the dimerization of the carboxyl groups with its associated hydrogen bonding. The difference between the C-O bond distances is 0.047 angstroms. (1.271 versus 1.224 angstroms) The conditions around the oxygen atoms are therefore quite similar. The question then arises of the possibility of tautomerism of the protons in the dimerized carboxyl groups. A difference Fourier map shows the presence of the electron in the position associated with the longer C-O bond length, but none in the second position. At the present level of refinement of the structure, we can say that probably no more than 20% tautomerism occurs.

The carboxyl groups are tilted out of the plane of the benzene ring with the oxygen atoms being 0.12 angstroms above and below the plane respectively. The amine group and the benzene ring are co-planar to within the accuracy of these calculations.

The carbon to carbon distance from the benzene to the carboxyl group compares favorably to the B structure of the para-aminobenzoic acid (Ref. 1) and to that of 2-amino-3-methylbenzoic acid. (Ref. 4) The carbon to nitrogen distance is larger than in the other two cases cited probably due to the absence of resonance forms resulting from the protonation of the amine group. It is also noted that several of the carbon to carbon bond distances are found to be smaller than would normally be expected for meta- substitution. The differences in the carbon to carbon bond distances in the benzene ring are about 0.04 angstroms; this difference exceeds the standard deviation in these values and is even greater than the differences found in the para- and the 2-amino-3-methyl-. Our knowledge of this effect is inconclusive.

REFERENCES

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Bozeman, Montana, paper B-14, p. 29 (1964)
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19, 797 (1965)
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Fourier Map of Benzene Ring

Fig. 1

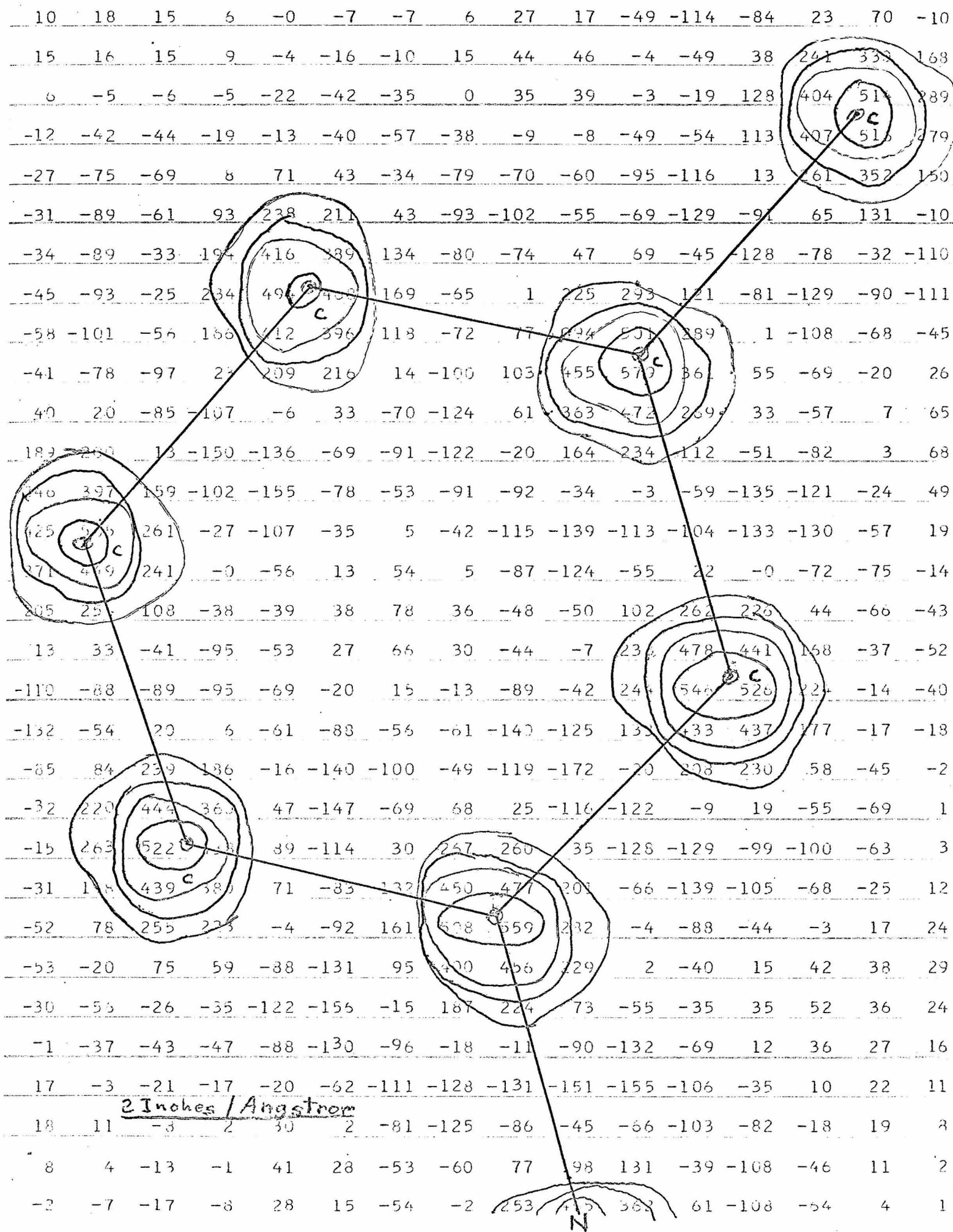


Fig. 2

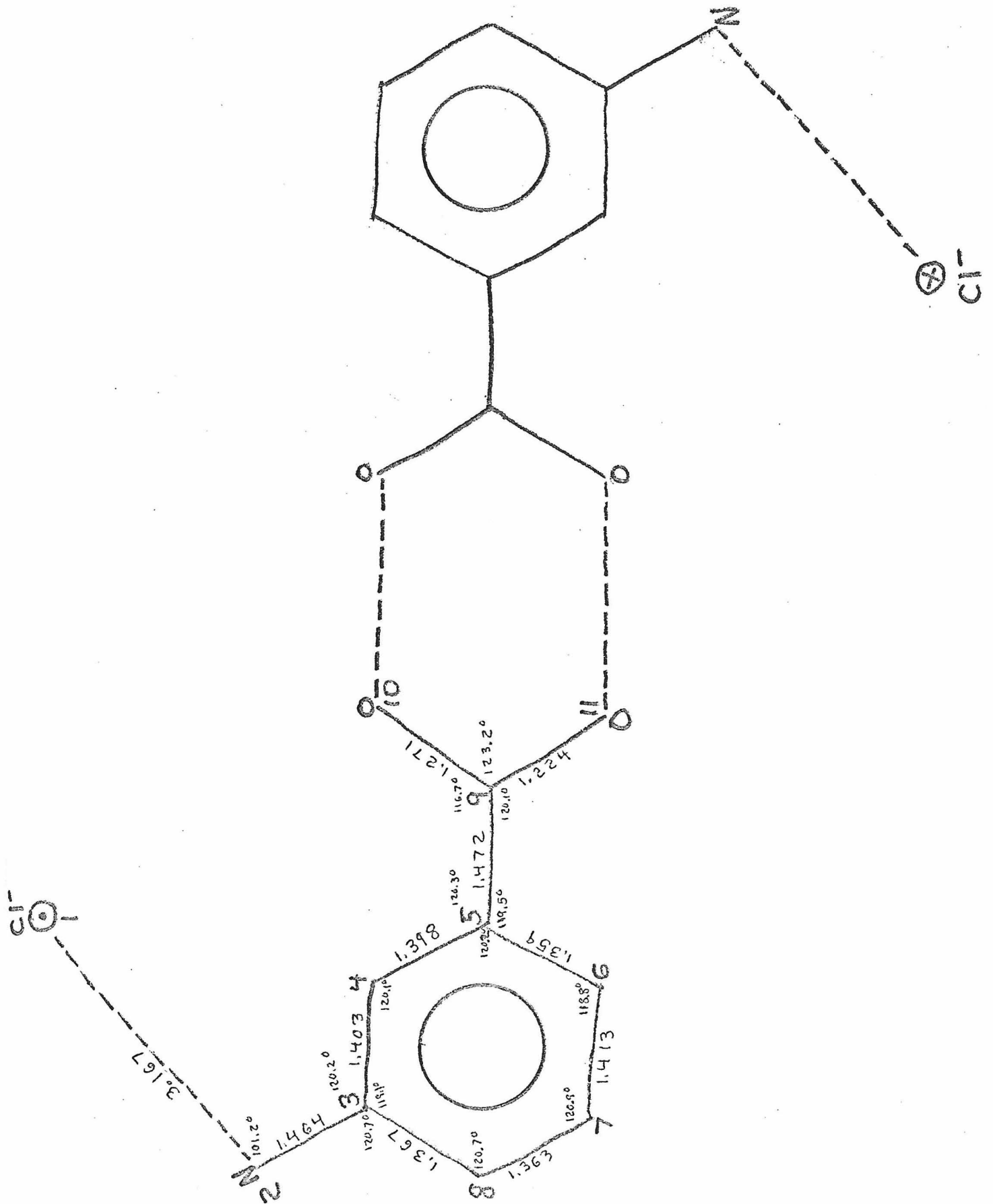


Table 1

BOND DISTANCES

<u>Atom</u>	<u>Atom</u>	<u>Distance</u>
1	2	3.167
2	3	1.464
3	4	1.403
3	8	1.367
4	5	1.398
5	6	1.359
5	9	1.472
6	7	1.413
7	8	1.363
9	10	1.271
9	11	1.224

Table 2

BOND ANGLES

<u>Atom</u>	<u>Atom</u>	<u>Atom</u>	<u>Angle</u>
1	2	3	101.2
2	3	4	120.2
2	3	8	120.7
8	3	4	119.1
3	4	5	120.1
4	5	6	120.2
4	5	9	120.3
6	5	9	119.5
5	6	7	118.8
6	7	8	120.9
7	8	3	120.7
5	9	10	116.7
5	9	11	120.1
10	9	11	123.2

Table 3

The Heavy-Atom ParametersAnd Their Standard Deviations (in parentheses)

<u>COORDINATES</u>			
<u>Atom</u>	<u>x</u>	<u>y</u>	<u>z</u>
Cl- (1)	.1436(4)	.2747(3)	.0668(1)
N (2)	.4077(13)	.8073(9)	.0923(4)
C (3)	.5529(15)	.8352(11)	.1835(5)
C (4)	.5038(15)	1.0240(11)	.2506(5)
C (5)	.6315(15)	1.0442(11)	.3389(5)
C (6)	.8154(15)	.8875(13)	.3590(5)
C (7)	.8646(16)	.6996(12)	.2904(5)
C (8)	.7328(17)	.6756(11)	.2050(5)
C (9)	.5749(17)	1.2391(12)	.4102(5)
O (10)	.3789(12)	1.3662(8)	.3907(3)
O (11)	.7176(12)	1.2750(8)	.4831(3)

Atoms are labeled on Fig. 2

<u>ANISOTROPIC TEMPERATURE FACTOR</u>						
<u>Atom</u>	<u>b₁₁</u>	<u>b₂₂</u>	<u>b₃₃</u>	<u>b₁₂</u>	<u>b₁₃</u>	<u>b₂₃</u>
Cl- (1)	567(13)	229(6)	39(1)	275(12)	-67(5)	-51(3)
N (2)	603(38)	248(19)	34(3)	273(43)	-36(17)	-34(11)
C (3)	513(49)	227(23)	34(4)	64(52)	38(21)	-33(14)
C (4)	486(44)	198(22)	41(4)	218(48)	-19(20)	24(14)
C (5)	405(42)	224(22)	36(4)	161(47)	-4(19)	-8(14)
C (6)	397(43)	342(26)	34(4)	190(55)	-19(19)	4(15)
C (7)	633(53)	313(28)	41(4)	358(61)	19(24)	3(17)
C (8)	611(50)	208(23)	46(4)	247(54)	35(24)	-25(15)
C (9)	508(50)	278(26)	31(4)	118(56)	13(21)	-33(16)
O (10)	737(38)	359(20)	43(3)	441(45)	-49(17)	-89(12)
O (11)	755(36)	404(21)	44(3)	391(45)	-129(17)	-94(13)

expressed in the form $\exp(-(b_{11}h^2 + b_{22}k^2 + b_{33}l^2 + b_{12}hk + b_{13}hl + b_{23}kl))$

all anisotropic temperature factors are multiplied by 10^4

Table 4

Observed and Calculated Structure Factors

Within each group are given the values of $1, 100 F(\text{obs.})$, and $100 F(\text{calc.})$. Reflections indicated with an asterisk were omitted from the least-squares refinement. A minus sign preceding $F(\text{obs.})$ signifies 'less than'; these reflections have been omitted from the refinement unless the value $F(\text{calc.})$ exceeds the threshold value of $F(\text{obs.})$.

0 0 L
0 ----* 8235

-5 2 L
0 409 259
1 533 489
2 355 210
3 529 555
4 -259 181
5 -247 -57
6 689 660
7 -209 -115
8 459 -404
9 276 -288

-5 3 L
0 330 186
1 206 167
2 -247 18
3 275 -144
4 344 -291
5 340 -364
6 318 -218

-5 4 L
0 560 -467
1 339 -276
2 379 -325
3 432 -427
4 410 -383

-4 2 L
0 678 573
1 729 667
2 834 697
3 -270 171
4 630 550
5 670 -600
6 1526 1416
7 443 338
8 486 -532

-4 3 L
0 -290 -203
1 -292 37
2 829 -531
3 1062 -953
4 661 490
5 851 -656
6 815 -713
7 726 -333
8 482 75
9 -247 356
10 394 282
11 238 377
12 283 149

-4 4 L
0 647 -592
1 763 -577
2 552 -392
3 -231 -60
4 -243* 237
5 687 625
6 614 603
7 407 310
8 510 412
9 182 -177

-3 1 L
0 -72 63
1 302 -176
2 1913 2193
3 649 807
4 1304 1414
5 254 302
6 458 525
7 407 -403
8 711 -793
9 763 -866
10 315 -294
11 665 -659
12 404 -433
13 -197 -99
14 268 187

-3 2 L
0 437 191
1 690 691
2 807 -861
3 542 -563
4 724 -811
5 -155 -17
6 493 -500
7 817 -899
8 -192 -123
9 -209 -273
10 252 211
11 453 489
12 -198 -79
13 -178 210
14 -150* 463
15 309 194

-3 3 L
0 529 -422
1 1240 1098
2 528 -379
3 1563 1294
4 2335 2209
5 1466 1244
6 -195 158
7 604 528
8 747 813
9 650 427
10 871 738
11 830 659

-3 4 L
0 527 -412
1 603 -596
2 1893 1719
3 220 195
4 352 205
5 386 384
6 792 639
7 663 643
8 -200 143
9 -186 30
10 -192* -348
11 350 -333
12 167 -174

-3 5 L
0 683 555
1 922 881
2 733 708
3 785 721
4 519 406
5 458 -343
6 315 -320
7 1296 1108
8 170 -215
9 472 -472

-3 6 L
0 244 -177
1 -168 79
2 359 -271
3 434 -475
4 807 -753

-2 1 L
0 3941 4127
1 2303 2165
2 795 752
3 582 636
4 825 756
5 1755 1851
6 1841 1997
7 510 -302
8 1020 -992
9 706 -555
10 706 -649
11 -185 28
12 470 392
13 728 647
14 395 408
15 -189 81
16 -160 143
17 163 224

-2 2 L
0 1607 1713
1 1039 1109
2 185 -1
3 1543 1453
4 925 -671
5 1515 1348
6 1073 -924
7 817 -819
8 876 878
9 -181 24
10 779 635
11 798 782
12 966 905
13 403 446
14 312 283
15 189 49

-2 3 L
0 3069 2659
1 1659 1401
2 856 -762
3 1110 1113
4 851 777
5 506 472
6 1365 1261
7 1058 1033
8 814 787
9 750 743
10 585 -450
11 488 -419
12 -295 15
13 486 -481
14 452 -429
15 214 -283

-2 4 L
0 942 881
1 295 203
2 1031 -999
3 2683 2616
4 827 939
5 527 549
6 366 439
7 356 -310
8 345 -248
9 -204* -181
10 901 1003
11 736 -594
12 800 -911
13 220 -217

-2	5	L
0	2046	2135
1	-186	61
2	395	-405
3	599	-588
4	511	-519
5	526	-507
6	676	-772
7	626	-640
8	544	-513
9	-168	-250
10	419	358
11	270	317

-2	6	L
0	438	-359
1	402	-433
2	452	-466
3	810	-846
4	449	-490
5	777	-809
6	432	339
7	340	364

-1	1	L
0	973	1059
1	624	552
2	2680	2474
3	477	-520
4	3491	3701
5	1999	1782
6	2457	2350
7	1495	1415
8	-149	40
9	698	-705
10	881	794
11	1216	1138
12	1030	1022
13	850	817
14	856	745
15	751	641
16	-180	186
17	-148	-130

-1	2	L
0	2433	2389
1	2601	2488
2	733	-674
3	3445	3711
4	2124	2223
5	927	897
6	594	533
7	1258	1311
8	920	890
9	1732	1510
10	1281	1261
11	861	798
12	-186	183
13	581	-581
14	617	-684
15	372	-360
16	537	-646

-1	3	L
0	373	405
1	-136	188
2	1289	1197
3	740	747
4	1354	1256
5	489	443
6	1895	1885
7	197	-117
8	244	184
9	131	-68
10	-186	-51
11	793	-680
12	657	-573
13	612	-607
14	141	-157
15	830	-922

-1	4	L
0	1217	1201
1	2568	2523
2	1253	1287
3	335	308
4	371	368
5	-181	-49
6	818	-785
7	759	-742
8	1375	1638
9	-250*	1189
10	-189	-134
11	161*	-252
12	-176*	71
13	623	456
14	561	639

-1	5	L
0	1420	1455
1	1002	1052
2	500	-391
3	872	-999
4	685	-753
5	773	-909
6	411	-425
7	172	210
8	293	-303
9	299	356
10	-162	42
11	553	582

-1	6	L
0	1038	1219
1	842	-889
2	326	-448
3	116	-30
4	-185	183
5	381	369
6	444	559
7	536	591

0	0	L
2	629	748
3	850	963
4	1176	1266
5	902	-786
6	2574	2111
7	3822	3284
8	1159	1116
9	1879	1828
10	789	414
11	-234*	224
12	911	659
13	-238	186
14	400	461
15	717	696
16	207	201
17	405	459

0	1	L
-17	328	350
-16	-235*	256
-15	223	-156
-14	867	-878
-13	428	-390
-12	1192	1173
-11	521	-576
-10	1634	1611
-9	1238	1077
-8	850	635
-7	986	-890
-6	1317	1336
-5	2182	2295
-4	728	742
-3	3719	5410
-2	2552	2977
-1	903	-907
0	1514	1402
1	1106	1053
2	401	-90
3	3305	3374
4	1070	-883
5	2761	2832
6	721	-614
7	280	-251
8	-188*	2
9	839	619
10	-210	-80
11	1496	1348
12	2768	2654
13	709	-625
14	594	609
15	-295*	-233
16	-228	-307
17	-239*	-212
18	488	-548

0	2	L
-17	263	-456
-16	-254*	-37
-15	-255*	-240
-14	603	-753
-13	542	-489
-12	-234*	100
-11	-225	149
-10	-213	-3
-9	-219*	1517
-8	1722	1768
-7	1771	1641
-6	402	249
-5	1434	1215
-4	-113*	48
-3	-104*	2300
-2	2079	2893
-1	2417	2252
0	4706	4501
1	855	-874
2	2582	2865
3	3689	4119
4	-141	291
5	1029	1216
6	2329	2538
7	1126	1115
8	1358	1445
9	1414	1416
10	2231	2033
11	-237	75
12	-244*	-358
13	-231	-297
14	656	-619
15	567	-687
16	227	-222

0	3	L
-17	388	-325
-16	-238*	258
-15	-264*	234
-14	660	548
-13	684	596
-12	588	478
-11	-227	62
-10	640	482
-9	-202*	-79
-8	-186	94
-7	958	-919
-6	1203	1214
-5	1195	1294
-4	432	262
-3	1172	1298
-2	1073	1110
-1	716	-658
0	1789	1846
1	545	-554
2	702	566
3	2882	2523
4	1628	1333
5	1162	1037
6	-192	116
7	-207*	121
8	368	-156
9	1085	-859
10	1213	-990
11	1389	1269
12	1050	1011
13	-234	-80
14	313	-227
15	936	855

0	4	L
-16	222	194
-15	-224	185
-14	-234	88
-13	577	546
-12	393	358
-11	1032	-994
-10	745	-744
-9	560	-379
-8	973	-937
-7	301	-333
-6	613	-545
-5	683	-644
-4	1436	1368
-3	1015	1057
-2	1690	1525
-1	1316	1257
0	1161	1018
1	-122	73
2	-143	53
3	1080	1042
4	-178	-209
5	2010	1600
6	1218	1142
7	774	-699
8	-237	-105
9	403	-293
10	306	299
11	356	319
12	958	900
13	369	-185
14	522	600

0	5	L
-14	900	-747
-13	-225	-37
-12	-234	225
-11	427	-359
-10	-232	-108
-9	-233	73
-8	-237	77
-7	898	727
-6	527	360
-5	907	736
-4	290	205
-3	236	168
-2	236	128
-1	917	698
0	748	-548
1	1297	1260
2	978	-872
3	842	-647
4	769	-689
5	-234	-176
6	652	-441
7	-235	191
8	1049	985
9	796	803
10	591	637
11	636	688

0	6	L
-11	440	269
-10	533	427
-9	540	463
-8	338	269
-7	-235	14
-6	266	146
-5	313	-214
-4	274	-202
-3	1069	-907
-2	1387	1427
-1	427	306
0	-235	-73
1	379	-297
2	343	208
3	373	303
4	728	632
5	938	915
6	315	302
7	393	371

0	7	L
-8	284	-244
-7	275	-261
-6	296	-250
-5	391	-386
-4	270	249
-3	334	-260
-2	286	-253
-1	207	169
0	473	401
1	573	545
2	511	530
3	437	323

1	-6	L
1	942	-992
2	391	432
3	425	-517
4	124	139
5	347	351
6	295	325
7	888	893
8	539	450

1	-5	L
1	-185	-66
2	434	1022
3	1030	737
4	1066	1038
5	419	304
6	433	417
7	373	302
8	-185	-3
9	-183	-226
10	552	-435
11	573	-560
12	640	-665
13	270	-222

1	-4	L
1	615	633
2	416	-418
3	281	296
4	1062	1172
5	157	-203
6	1223	1357
7	842	-858
8	620	-633
9	592	-587
10	606	674
11	627	598
12	685	-732
13	724	610
14	474	385
15	344	257
16	272	264

1	-3	L
1	1331	1335
2	198	129
3	3675	3278
4	1987	1781
5	299	-514
6	870	-869
7	607	626
8	500	279
9	123	-156
10	1941	1616
11	1578	1426
12	425	338
13	-184	135
14	249	-119
15	-188	-95
16	-261*	-420
17	278	-244
18	173	-414
19	405	-137

1	-2	L
1	868	-739
2	2455	2443
3	570	645
4	1329	1158
5	761	712
6	2192	2250
7	305	390
8	895	920
9	434	-395
10	171	-151
11	-232*	-331
12	227	-192
13	705	-669
14	-183	-78
15	880	-889

1	-1	L
1	2293	2125
2	2835	3195
3	4461	4313
4	373	491
5	2029	1946
6	481	415
7	131	-197
8	2320	2478
9	1720	1625
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3	326	-293
4	-243	-284
5	-230	-115
6	-213	-14
7	291	363
8	206	278

4	4	L
-8	434	479
-7	-187	-95
-6	257	-297
0	-214	-74
1	-208	-1
2	281	218
3	-186	312
4	-169	354
5	-142	-44
6	74	213

5	-4	L
1	-159	-15
2	-157	113
3	345	279
4	370	239
5	431	367
6	234	318

5	-3	L
1	-256	112
2	747	602
3	709	570
4	-240	64
5	-228	131
6	-211	59
7	-188	-168
8	-151	-590
9	549	-310

5	-2	L
1	531	-457
2	-277	19
3	460	-439
4	-263	-290
5	431	-360
6	274	-230

5	-1	L
1	-293	-326
2	466	-378
3	540	-431
4	338	-422
5	-279	230
6	510	477
7	-243	39
8	415	504
9	278	300

5	0	L
-9	202	303
-8	-201	-46
-7	-221	-92
-6	273	-196
-5	248	-218
-4	-245	-267
-3	-247	-100
-2	218	-304
-1	346	-275
0	-249	101
1	-248	31
2	532	442
3	666	585
4	604	-454
5	514	546
6	378	344

5	1	L
-4	289	-237
-3	187	315
-2	328	107
-1	-227	158
0	447	420
1	-225	11
2	628	582
3	337	278
4	-206	-187
5	248	-171
6	265	-290
7	191	-323

5	3	L
0	-142	-361
1	328	-123